MARIS STELLA COLLEGE (AUTONOMOUS), VIJAYAWADA

A College with Potential for Excellence

NAAC Accredited & ISO 21001: 2018 Certified



PROGRAMME REGISTER: 2023-26
DEPARTMENT OF CHEMISTRY

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PROGRAMME OUTCOMES

(POs) 2023-26

Students of all Undergraduate Programmes at the time of graduation will be able to possess

PO1: Essential Knowledge:

Have comprehensive discipline knowledge and understanding, the ability to engage with different schools of thought and to apply their knowledge in practice including in multidisciplinary or multi-professional contexts.

PO2: Creative, Critical Thinking and Problem-Solving Abilities:

Be effective problem-solvers, able to apply critical and evidence-based thinking to conceive innovative responses to future challenges.

PO3: Teamwork and Communication Skills:

Convey ideas and information effectively to a range of audiences for a variety of purposes and contribute in a positive and collaborative manner to achieving common goals.

PO4: Motivated, Self-directed, and Life-long Learning:

Exhibit life-long skills; broad-based multiple career oriented general skills; self and field-based learning skills; digital skills; preparedness for living, learning and working in any environment.

PO5: Professionalism and Leadership Readiness:

Engage in professional behaviour and have the potential to be entrepreneurial and take leadership roles in their chosen occupations and communities.

PO6: Intercultural and Ethical Competency:

Be responsible and effective global citizens whose personal values and practices are consistent with their roles as responsible members of society.

PO7: Self-awareness and Emotional Intelligence:

Be self-aware and reflective, flexible and resilient and act with integrity and take responsibility for their actions as empowered women.

PO8: Social Responsibility and Effective Citizenship:

Exhibit social responsibility and compassionate commitment; Be sensitive to and demonstrate institution in matters of environment, gender and other social issues to promote an equitable society and sustainable development.

PROGRAMME SPECIFIC OUTCOMES

(PSOs) 2023-26

At the end of the programme students will be able to possess/exhibit:

PSO1: Quantitative Analysis:

Interpret principles, classifications, concepts, theories and mechanisms learnt.

PSO2: Practical and Analytical Skills:

Analyse hypotheses, procedures, properties, experimental facts and draw conclusions.

PSO3: Logical and Critical Thinking:

Apply knowledge and techniques in sample analysis, problem-solving, results, and production.

PSO4: Teamwork and Communication:

Develop communicative competence, creative and critical thinking, practical, technical and employability skills, social sensibility and responsibility.

Course Outcomes

(COs) 2023-26

S.No.	Sem	Course Code	Course Title	Course Outcomes (COs)				
1	Ι	23SCCCEA14	Essentials and Applications of	CO1: Apply mathematical principles to solve various problems across complex numbers, trigonometry, vectors, and statistical analysis involving data sets.				
			Mathematical, Physical & Chemical Sciences	CO2: Summarize key physics principles, including measurements, motion, thermodynamics, wave behaviour, electromagnetism, atomic structure, and theories of the universe.				
				CO3: Outline the fundamental concepts of chemistry and their relevance in daily life.				
				CO4: Elaborate the interconnectedness of math, physics and chemistry and explain how they predict phenomena in diverse contexts.				
				CO5: Discuss about computer evolution, including the internet, network types and understand ethical issues in network security, cryptography, privacy and data protection.				
2	Ι	23SCCCAS14	Advances in Mathematical	CO1: Identify the applications of mathematics in physics and chemistry to solve real-world problems.				
			Physical & Chemical Sciences	Chemical	Chemical Sciences CO2: Explain renewable energy general energy-efficient materials and recent administration nanotechnology, biophysics, medical planterials science. CO3: Outline computer-aided drug described chemical biology, the impact of chemical biology.	Chemical	Chemical Sciences energy nanote	CO2: Explain renewable energy generation, storage, energy-efficient materials and recent advancements in nanotechnology, biophysics, medical physics, and materials science.
						CO3: Outline computer-aided drug design, Nano sensors, chemical biology, the impact of chemical pollutants on ecosystems and human health, and methods for dye removal using catalysis.		
				CO4: Elaborate the interconnectedness of math, physics and chemistry and apply these principles to explain phenomena in diverse contexts.				
				CO5: Summarize the advanced computer science topics, such as number systems, signals, error detection and correction, multiplexing, transmission media, and networking devices.				
3	Ι	23MDCCS12	Principles of Chemical Sciences	CO1: Outline the classification, principles, theories and nuclear reactions of atoms and molecules				

				CO2: Classify and study the bonding, periodic properties of elements.
				CO3: Explain the types and properties of salts, importance of chemistry in daily life.
4	II	23CHCCGI23	General & Inorganic Chemistry	CO1: Outline the structure of atom, electronic configuration, classification of elements and periodic properties
				CO2: Explain properties, lattice energy and stability of ionic compounds.
				CO3: Summarize the concepts, theories of bonding and molecular structures
				CO4: Understand theories of acids and bases and types of chemical reactions
				CO5: Solve concept-based problems
5	II	23CHP1AS21	Qualitative Analysis of	CO1: Analyze simple salt by adapting systematic procedure
			Simple Salt - Practical	CO2: Apply the concepts of common ion effect, solubility product to qualitative analysis
				CO3: Use glassware, equipment, chemicals and follow experimental procedures in the laboratory.
6	II	23CHCCIC23	Inorganic Chemistry I	CO1: Outline classification, preparations and molecular structures of listed compounds of p-block elements
				CO2: Elaborate the characteristic properties of d and f – block elements.
				CO3: Explain the fundamental concepts of radioactivity, nuclear reactions and applications
				CO4: Solve concept-based problems
7	II	23CHP2IC21	Inorganic Chemistry I -	CO1: Prepare the listed inorganic compounds applying standard procedures.
			Practical	CO2: Determine melting point of the given compound.
				CO3: Use glassware, equipment, chemicals and follow experimental procedures in the laboratory.
8	III	23CHCCFO33	Fundamentals in Organic Chemistry	CO1: Explain the organic reagents, intermediates, electron displacement concepts and their applications.

				CO2: Outline the preparations, physical and chemical properties of alkanes, cycloalkanes, alkenes, alkynes
				CO3: Summarize the mechanisms of certain chemical reactions
				CO4: Discuss the concept of aromaticity, orientation of substitution with relevant examples.
				CO5: Solve concept-based problems
9	III	23CHP3OA31	Organic Qualitative Analysis - Practical	CO1: Adapt systematic procedure and perform organic compound analysis to identify the organic functional group and name of the compound. CO2: Determine the boiling/melting point of the given organic compound.
				CO3: Use glassware, equipment, chemicals and follow experimental procedures in the laboratory.
10	III	23CHCCOC33	Organic Chemistry	CO1: Outline the preparations and properties of alkyl, aryl halides, alcohols and phenols.
			(Halogens & Oxygen Organic Compounds)	CO2: Summarize the preparative methods and characteristic reactions of carbonyl compounds, carboxylic acids and active methylene compounds. CO3: Discuss the molecular structure, physical and chemical properties of carbohydrates.
				CO4: Write the listed mechanisms and named reactions
				CO5: Solve concept-based problem
11	III	23CHP4OP31	Organic Preparations - Practical	CO1: Perform organic synthesis for the compounds mentioned in the syllabus. CO2: Execute common laboratory techniques including reflux, distillation, recrystallization, vacuum filtration. CO3: Use glassware, equipment, chemicals and follow experimental procedures in the laboratory.
12	III	23CHCCPC33	Physical Chemistry I (Solutions & Electrochemistr y)	CO1: Elucidate the concepts and laws pertaining to behaviour of liquid solutions. CO2: Outline the colligative properties and determination methods of molar mass. CO3: Explain various photo processes and the laws of photochemistry.
				CO4: Summarize the electrochemical concepts and their applications in electro analytical techniques.
				CO5: Solve concept-based problems

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13	III	23CHP5PC31	Physical	CO1: Determine CST of Phenol-water system and examine
			Chemistry I - Practical	the effect of electrolyte on CST.
			Fractical	CO2: Perform potentiometric and conductometric titrations
				for quantitative estimations.
				CO3: Use glassware, equipment, chemicals and follow
				experimental procedures in the laboratory.
14	III	23CHCCIP33	Inorganic &	CO1: Summarize the theories of bonding and isomerism of
			Physical	complex compounds.
			Chemistry	CO2: Elaborate the inorganic reaction mechanism and
				stability of metal concepts.
				CO3: Outline the classification of organometallic
				compounds and discuss the metal carbonyls.
				CO4: Explain the concepts and laws of thermodynamics
				and deduce thermodynamic relations.
				CO5: Solve concept-based problems
15	III	23CHP6IA31	Qualitative	CO1: Analyze Mixture salt by adapting systematic
			Inorganic	procedure
			Analysis - Practical	CO2: Apply the concepts of common ion effect, solubility
			Tractical	product to qualitative analysis
				CO3: Use glassware, equipment, chemicals and follow
				experimental procedures in the laboratory.
16	IV	23CHCCPC43	Physical	CO1: Summarize the theories of bonding and isomerism of
			Chemistry II	complex compounds.
			(States of	
			Matter, Phase Rule & Surface	CO2: Elaborate the inorganic reaction mechanism and
			Chemistry)	stability of metal concepts.
				CO3: Outline the classification of organometallic
				compounds and discuss the metal carbonyls.
				CO4: Apply phase rule to mentioned systems
				CO5: Solve concept-based problems.
17	IV	23CHP7PC41	Physical	CO1: Determine surface tension and viscosity of given
			Chemistry II -	liquids.
			Practical	CO2: Verify Freundlich adsorption isotherm.
				CO3: Use glassware, equipment, chemicals and follow
				experimental procedures in the laboratory.
18	IV	23CHCCGP43	General &	CO1: Draw molecular representations and explain the
10	1 4		Physical	concepts of optical isomerism.
			Chemistry	CO2: Outline the role of metals in biological systems.
				CO3: Describe the concepts of ionic equilibrium, common
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				ion effect and solubility product with applications.
				CO4: Deduce expression for rate constants, summarize
				theories of reaction rates and enzyme catalysis.
				CO5: Solve concept-based problems
19	IV	23CHP8VA41	Volumetric	CO1: Outline the theoretical and experimental aspects of
			Analysis -	volumetric analysis.
			Practical	CO2: Perform volumetric analysis for quantitative
				estimations.
				CO3: Use glassware, equipment, chemicals and follow
				experimental procedures in the laboratory.
20	IV	23CHCCOS43	Nitrogen	CO1: Elucidate the reaction mechanisms in Amines
			Containing	CO2: Outline the classification, preparations and properties
			Organic Compounds &	of amino acids, nitro compounds
			Spectroscopy	CO3: Discuss the molecular structure, physical and
				chemical properties of heterocyclic compounds.
				CO4: Use UV-Visible, IR spectral data to Interpret the
				molecular structures of simple molecules.
				CO5: Solve concept-based problems.
21	IV	23CHP9OS41	Organic	CO1: Prepare listed organic compounds
			Preparations &	CO2: Analyze IR spectra of given functional group
			IR Spectral Analysis -	compounds to interpret the molecular structure.
			Practical	CO3: Use glassware, equipment, chemicals and follow
				experimental procedures in the laboratory.

Mapping of COs with PSOs & POs

S.No.	Sem	Course Code	Course Title	COs	PSOs	POs
1	Ι	23SCCCEA14	Essentials and Applications of	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4,
			Mathematical, Physical &	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
			Chemical Sciences	CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO7, PO8
				CO4	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO7, PO8
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
2	I	23SCCCAS14	Advances in Mathematical	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
			Physical & Chemical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
			Sciences	CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO7, PO8
				CO4	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO7, PO8
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
3	I	23MDCCS12	Principles of Chemical	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
			Sciences	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO8
4	II	23CHCCGI23	General & Inorganic	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Chemistry	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
5	II	23CHP1AS21	Qualitative Analysis of	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Simple Salt - Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5

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				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
6	II	23CHCCIC23	Inorganic Chemistry I	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			3	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
7	II	23CHP2IC21	Inorganic Chemistry I -	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
8	III	23CHCCFO33	Fundamentals in Organic	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Chemistry	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
9	III	23CHP3OA31	Organic Qualitative	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Analysis - Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
10	III	23CHCCOC33	Organic Chemistry	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			(Halogens & Oxygen	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Organic Compounds)	CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Compounds)	CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
11	III	23CHP4OP31	Organic Preparations -	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
12	III	23CHCCPC33	Physical Chemistry I	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4

			(Solutions & Electrochemistr	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
			y)	CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	, ,
13	III	23CHP5PC31	Physical Chemistry I -	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
14	III	23CHCCIP33	Inorganic & Physical	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Chemistry	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
15	III	23CHP6IA31	Qualitative Inorganic	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Analysis - Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
16	IV	23CHCCPC43	Physical Chemistry II	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			(States of Matter, Phase	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Rule & Surface Chemistry)	CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
17	IV	23CHP7PC41	Physical Chemistry II -	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
18	IV	23CHCCGP43	General & Physical	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Chemistry	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4

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				CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
19	IV	23CHP8VA41	Volumetric Analysis -	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Practical	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
				CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
20	IV	23CHCCOS43	Nitrogen Containing	CO1	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Organic Compounds &	CO2	PSO1, PSO2, PSO3	PO1, PO2, PO4
			Spectroscopy	CO3	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO4	PSO1, PSO2, PSO3	PO1, PO2, PO4
				CO5	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4
21	IV	23CHP9OS41	Organic Preparations &	CO1	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			IR Spectral Analysis -	CO2	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5
			Practical	CO3	PSO1, PSO2, PSO3, PSO4	PO1, PO2, PO3, PO4, PO5

Mapping of Courses with PSOs

Course Title	PSO1 Quantitative Analysis	PSO2 Practical and Analytical Skills	PSO3 Logical, Critical Thinking	PSO4 Teamwork and Communication
Essentials and Applications of Mathematical, Physical & Chemical Sciences (EA)	V	V	~	~
Advances in Mathematical Physical & Chemical Sciences (AS)	V	V	V	~
Principles of Chemical Sciences (CS)	~	~	~	~
General & Inorganic Chemistry (GI)	V	~	v	~
Qualitative Analysis of Simple Salt – Practical (AS)	V	~	✓	~
Inorganic Chemistry I (IC)	V	~	v	~
Inorganic Chemistry I – Practical (IC)	V	~	v	~
Fundamentals in Organic Chemistry (FO)	~	~	~	~
Organic Qualitative Analysis – Practical (OA)	V	~	~	~
Organic Chemistry (Halogens & Oxygen Organic Compounds) (OC)	>	~	✓	'
Organic Preparations – Practical (OP)	V	·	•	✓
Physical Chemistry I (Solutions & Electrochemistry) (PC)	V	~	~	'
Physical Chemistry I – Practical (PC)	V	·	•	✓
Inorganic & Physical Chemistry (IP)	V	~	•	·
Qualitative Inorganic Analysis – Practical (IA)	V	~	~	'
Physical Chemistry II (States of Matter, Phase Rule & Surface Chemistry) (PC)	V	~	v	~
Physical Chemistry II – Practical (PC)	~	'	•	'
General & Physical Chemistry (GP)	~	~	~	~
Volumetric Analysis – Practical (VA)	v	~	~	~
Nitrogen Containing Organic Compounds & Spectroscopy (OS)	V	V	V	V

Organic Preparations & IR Spectral Analysis – Practical (OS)			
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Mapping of Courses with POs

Course	PO1 Essential Knowledge	PO2 Creative, Critical thinking and Problem- solving abilities	PO3 Teamwork and Communicatio n skills	PO4 Motivated, Self-directed and Life-long Learning	PO5 Professionalism and Leadership Readiness	PO6 Intercultural and Ethical Competency	PO7 Self-awareness and Emotional Intelligence	PO8 Social Responsibility and Effective Citizenship
EA	~	~	~	~			~	~
AS	V	V	V	~			~	~
CS	V	V	~	~				~
GI	~	~	~	~				
AS	~	V	~	~	V			
IC	~	~	~	·				
IC	v	~	•	•	•			
FO	/	~	•	~				
OA	/	~	•	~	~			
OC	~	~	•	~				
OP	~	~	~	~	•			
PC	~	~	~	~				
PC	~	~	~	~	•			
IP	~	~	~	~				
IA	/	'	•	~	•			
PC	/	'	•	~				
PC	/	'	•	~	•			
GP	V	V	~	V				
VA	V	V	~	~	V			
os	V	V	V	~				
os	V	V	~	V	~			